

Diastereoselective Reduction of (S)-1-Chloro-3-silyloxybutan-2-one. Synthesis of Enantiopure (2S,3R) and (2S,3S) O-tert-Butyldimethylsilyl-3,4-epoxybutan-2-ol.

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Abstract: (2S,3S)- And (2R,3S)-3-[(tert-butyldimethyl)silyloxy]-1-chlorobutan-2-ol have been obtained with high diastereoselectivity by reduction of enantiopure (S)-3-[(tert-butyldimethyl)silyloxy]-1-chlorobutan-2-one using different reducing agents. The chiral alcohols were transformed into the corresponding (2R,3S)- and (2S,3S)-3-[(tert-butyldimethyl)silyloxy]-1,2-epoxybutane. © 1999 Elsevier Science Ltd. All rights reserved.

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We have reported the direct preparation of chiral (S)-3-alkoxy-1-chlorobutan-2-one 1 by reaction of (chloromethyl)lithium generated in situ with O-protected ethyl lactate, [1].

In the present communication we report some synthetic applications of 1. Thus, reduction of 1 with different reducing agents led to both diastereoisomeric β -alkoxy alcohols syn-2 or anti-2 [2] with high diastereoselectivity; epoxidation of these diols provided both diastereoisomers of the 3-alkoxy-1,2-epoxides syn-3 and anti-3. The stereochemistry of the major diastereoisomers obtained in the reduction reactions was also established.

Scheme 1

Reduction using ketone 1a and NaBH₄, gave the desired O-monoprotected diol 2 in high yield, but no diastereoselectivity was observed. By contrast, reduction of 1b with NaBH₄ led to the anti-2 β -silyloxy alcohol with high diastereoselectivity. Since O-tert-butyldimethylsilylated ketone 1b shows enhanced diastereoselectivity relative to the corresponding O-benzylated

ketone 1a, we investigated the reduction with other reducing agents using 1b. Thus the reduction was also studied with NaBH₃CN, LiBH₄ and KBH₄ at -78 °C (Table 1). The major diastereoisomer obtained in these reductions was found to be *anti-2* in each case. In contrast, when 'Bu₂AlH or L-Selectride[®] in methanol were used, the *syn-2* product was synthesised as major diastereoisomer. (Table 1). The ratios of the diastereoisomers were obtained from the crude reaction mixtures using 'H-NMR and/or GC-MS. The mixtures of diastereoisomers obtained using NaBH₄ or L-Selectride[®] were subjected to flash column chromatography over silica gel (hexane:ethyl acetate, 20/1) to provide the pure diastereoisomer of *anti-2* or *syn-2*, respectively.

The stereochemistry of the reduction was determined by deprotection of the major diastereoisomer anti-2 or syn-2 with tetrabutylammonium chloride [3] and acetonization using 2,2-dimethoxypropane [4]. The absolute configuration of the 1,3-dioxolane cis-4 or trans-4 was established by NOESY experiments.

Synthesis of anti-2 (using NaBH₄) or syn-2 (using L-Selectride[®]) took place with no detectable racemization. The enantiomeric purity of anti-2 or syn-2 was determined by chiral GC (HP Chiral 20 B) analysis, which showed an enantiomeric excess (ee) > 99%; racemic mixtures of anti-2 or syn-2 were prepared to exclude the possibility of comigration of both enantiomers in the GC analysis.

Treatment of *anti-2* or *syn-2* with methyllithium at -78°C gave the corresponding alcoholate. When the reaction mixture was allowed to warm to room temperature the enantiopure alkoxy epoxides *anti-3* or *syn-3* were obtained, respectively.

Table 1: Reduction of chiral ketone 1b into anti-2 and syn-2.

Entry	Reduction agent	Yield (%) ^{b, c}	anti-2:syn-2
1	NaBH₄	86	84:16
2	NaBH₃CN	80	79:21
3	KBH₄	86	82:18

Entry	Reduction agent	Yield (%) ^{b, c}	anti-2:syn-2
4	LiBH ₄	73	66:34
5	L-Selectride®	75	5:95
6	ⁱ Bu₂AlH ^d	92	23:77

All reactions were carried out at -78 °C. All products were fully characterized by elemental analysis and spectroscopic methods (IR, 1'H- and 1'C-NMR, and HMRS). Isolated yield based on the starting ketone 1b. The reaction was performed in hexane.

In conclusion, the results reported represent a direct method for the preparation of enantiopure syn and anti O-silylated β, γ -epoxy alcohols, which are useful chiral building blocks in organic synthesis, with high yield and diastereoselectivity.

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